

Advanced Green Gas Technology (AGATE), phase 1

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Advanced Green Gas Technology (AGATE), phase 1

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AGATE1 partners, contacts & subjects

- ECN Luc Rabou Dry biomass gasification => methane
- RUG ST/OC Erik Heeres Wet biomass gasification => methane
- RUG CIO Sanne Palstra ¹⁴C analysis for “green” gas

Project finished December 2013

ECN and RUG ST/OC research continues in AGATE2

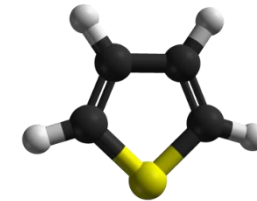
ECN R&D in AGATE₁

- Construction of pressurised test rig for conversion of organic sulfur
- Reforming of aromatic hydrocarbons
- Gas cleaning tests

$\text{HDS} = \text{HydroDeSulfurization}$

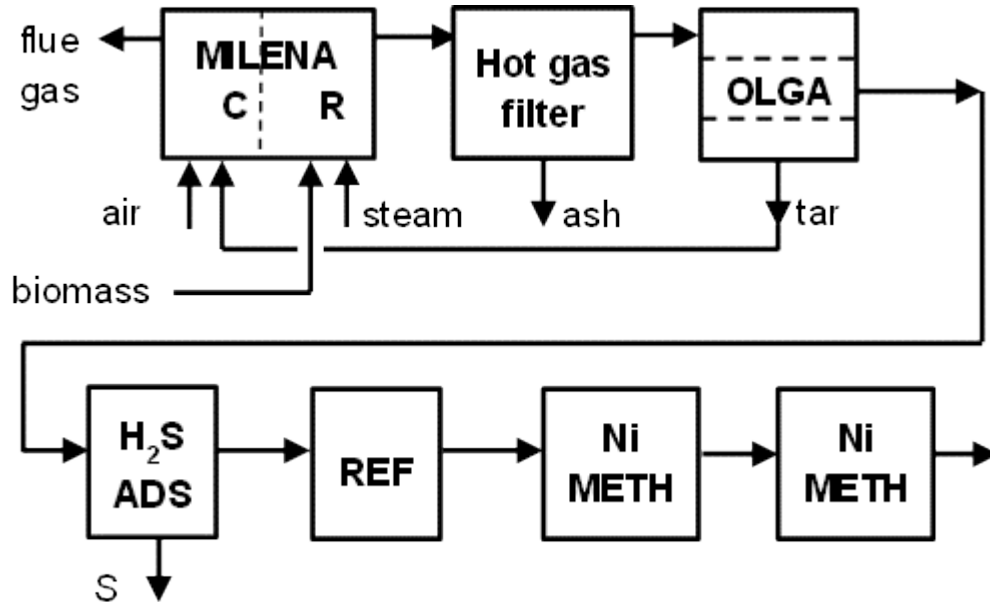
WHY?

Groen Gas 2.0
Milena-Olga-SNG



- Org-S (mainly $\text{C}_4\text{H}_4\text{S}$) poisons methanation catalyst
HDS slow at 1 bar; methanation requires high pressure
- Aromatic hydrocarbons => coke deposit, catalyst deactivation
- Assess performance

State of ECN R&D in 2010



- Gasifier (MILENA)
- Dust removal (filter)
- Tar removal (OLGA)
- H₂S removal (SACHA)
- C_xH_y reformer (SNG)
- Methanation (SNG)

At atmospheric pressure

ECN test rig in 2010



MILENA



SACHA

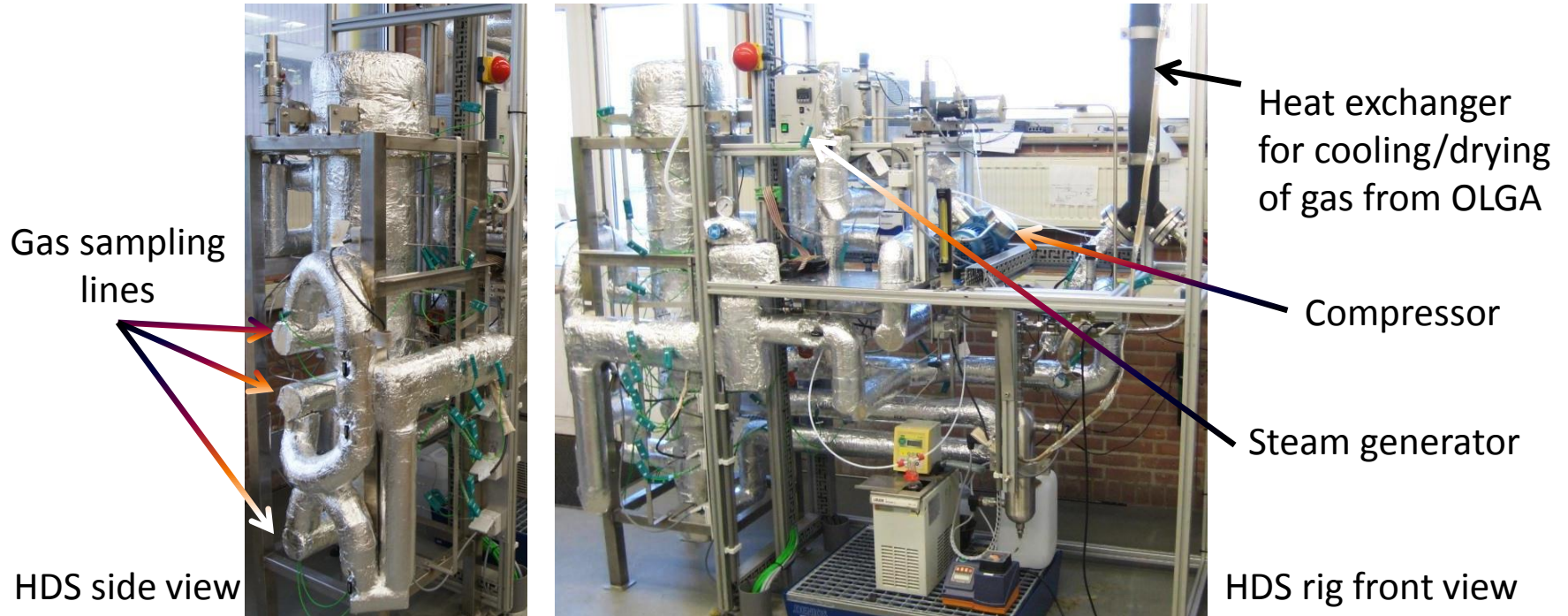


OLGA

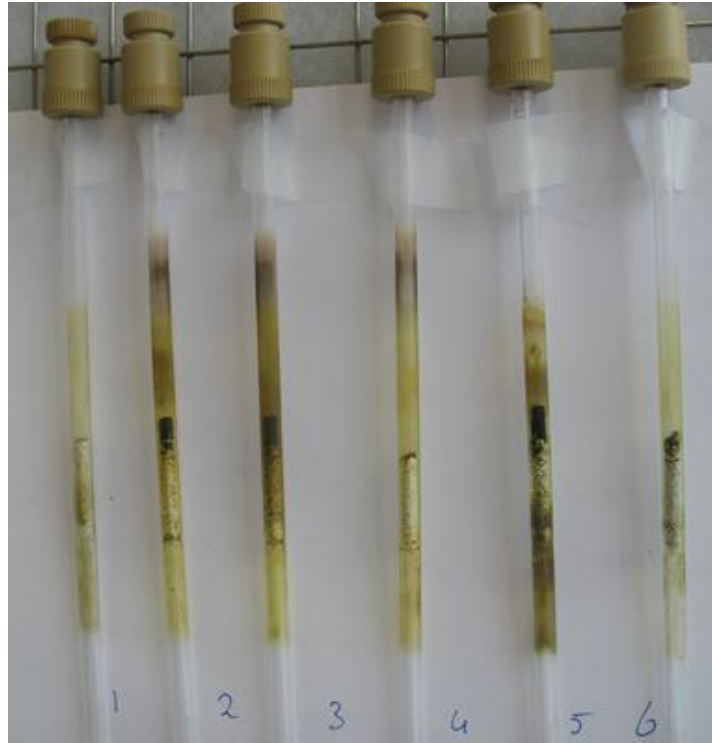


SNG

ECN pressurised HDS test rig



Benzene reforming



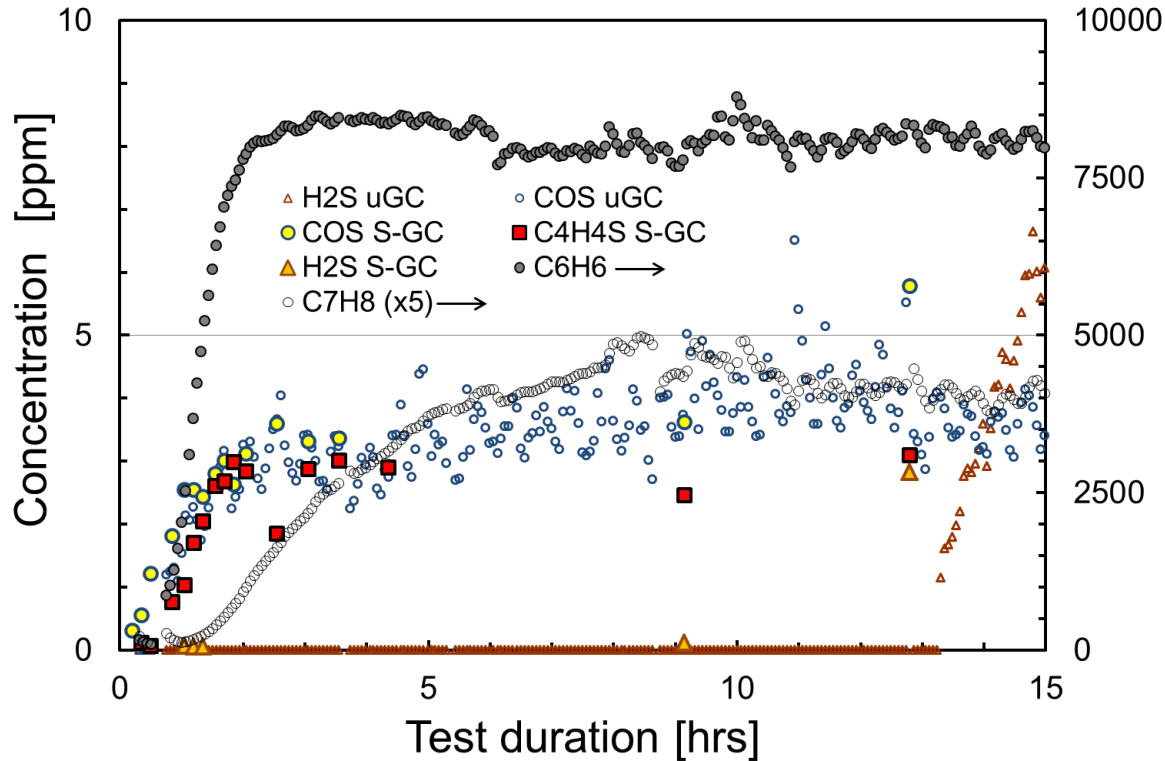
Microflow reactors with different catalysts:

amount of carbon deposit varies significantly

⇒ Catalyst selected for optimization of conditions

is continued in AGATE phase 2

Thiophene adsorption by active carbon



Impregnated
active carbon

Order of break-through:

COS

Benzene, thiophene

Toluene

H₂S

Conclusions of ECN research

Conversion of thiophene improves with pressure

HDS reactor size comparable to size of methanation reactors

Benzene reforming requires further research (in AGATE phase 2)

Thiophene adsorption possible, but has to compete with BTX

BTX + thiophene removal still an option (studied in SNG Impact)

RUG ST/OC R&D in AGATE₁

- Supercritical gasification in water (SCWG) batch experiments
- Construction test rig for continuous SCWG experiments
- SCWG continuous experiments

WHY and HOW?

- High conversion of wet biomass to CH₄
- Comparison of heterogeneous catalysts and catalyst nanoparticles
- From simple organic compound to more complex mixture

Batch SCWG: reactor



Tests at ~250 bar and 400°C

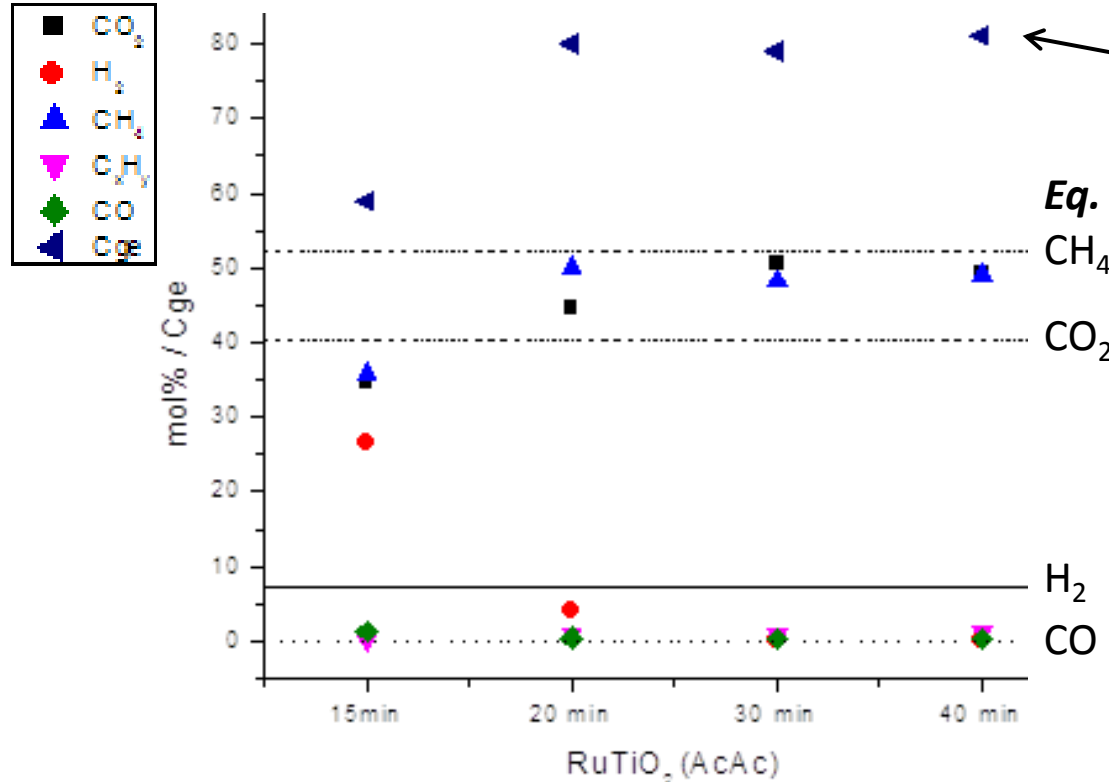
Volume 14 ml

Glycerol in water

Several Ru and Ni catalysts

commercial or home made
powder or nanoparticles

Batch SCWG: residence time

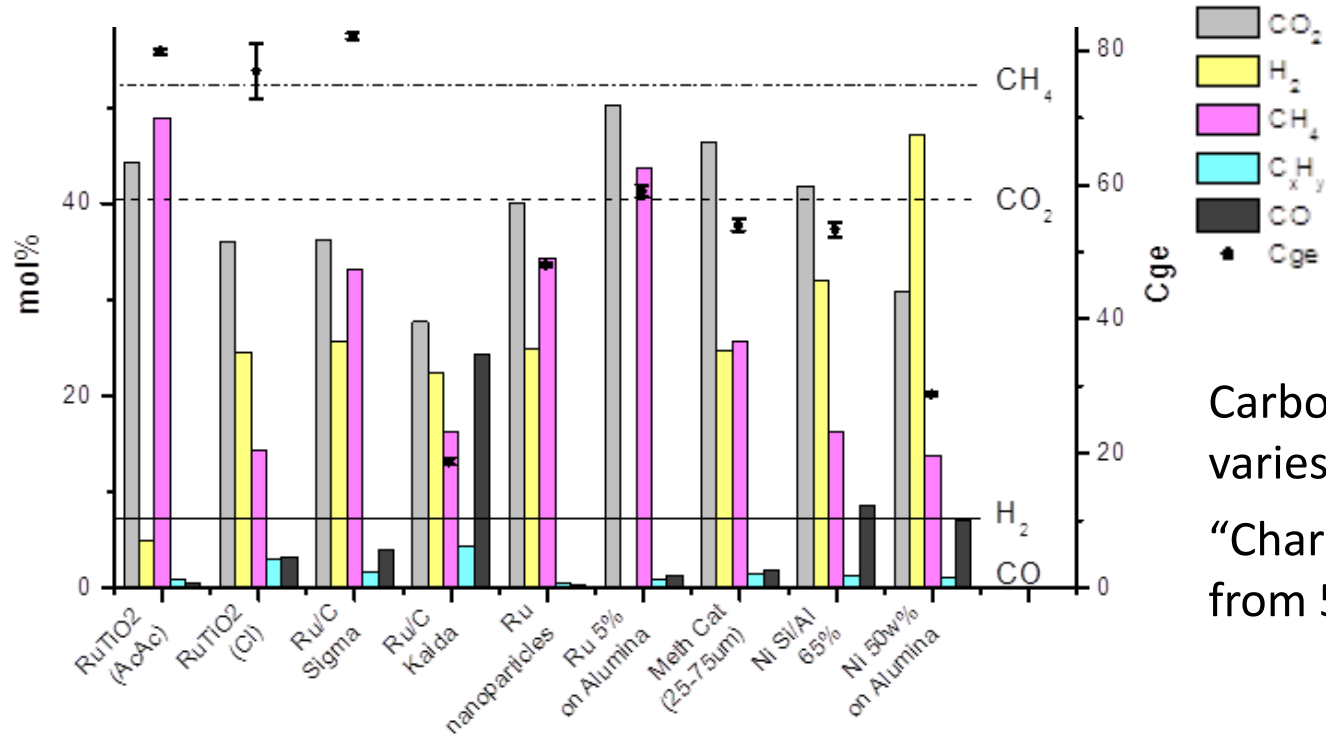


$$C_{ge} = \frac{\text{carbon in gas}}{\text{carbon in feed}}$$

Eq. CH₄
 Gas composition close to equilibrium in 20 to 30 minutes

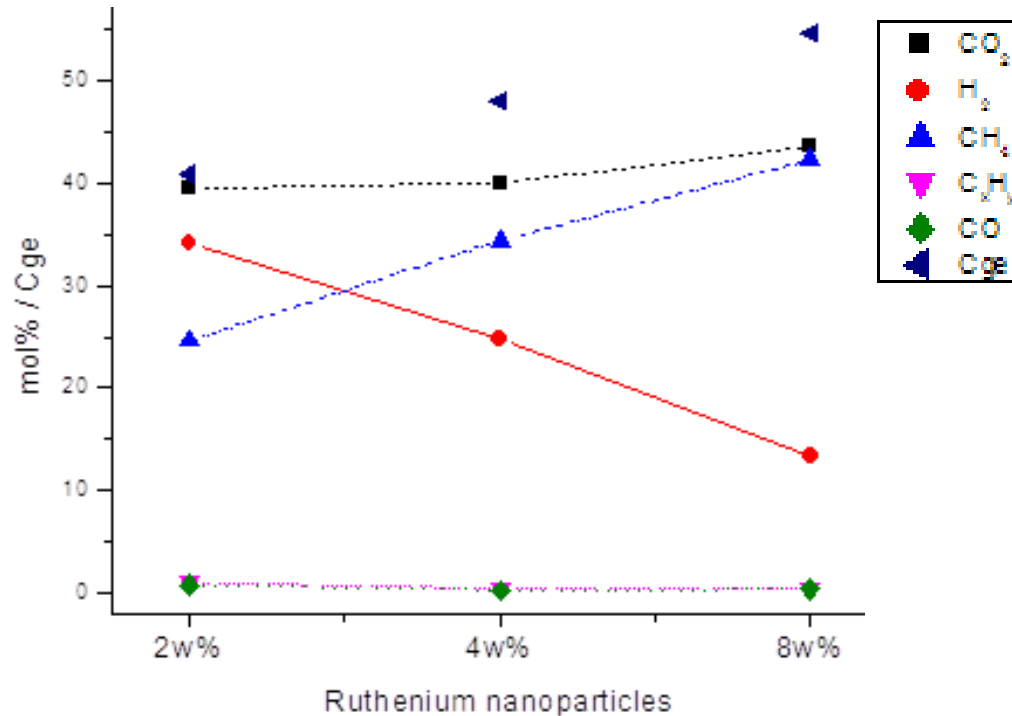
Maximum carbon to gas conversion in 20 minutes

Batch SCWG: gas composition & conversion



Carbon to gas efficiency varies from 19% to 82%
 “Char formation” varies from 5% to 33%

Batch SCWG nanoparticles



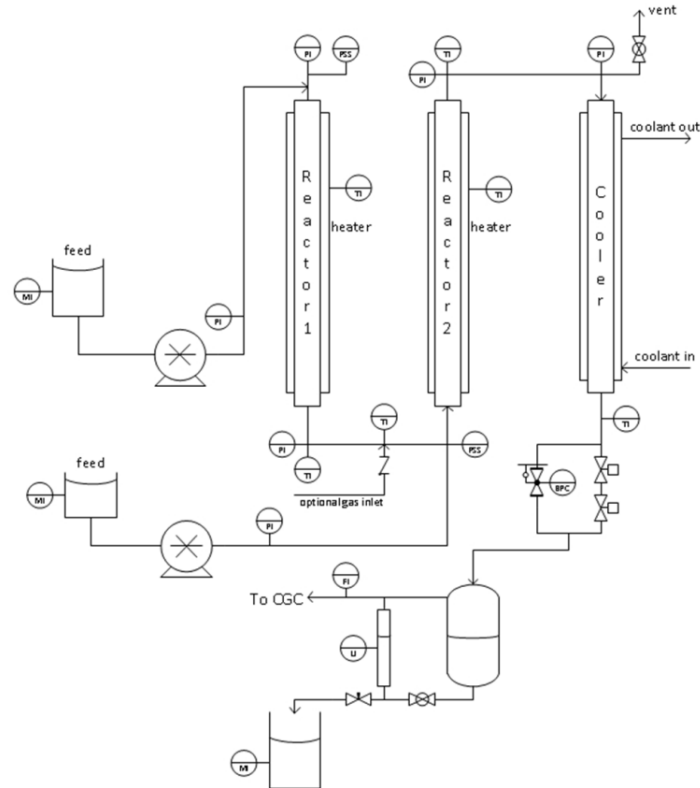
Limited conversion

Gas composition does not reach equilibrium

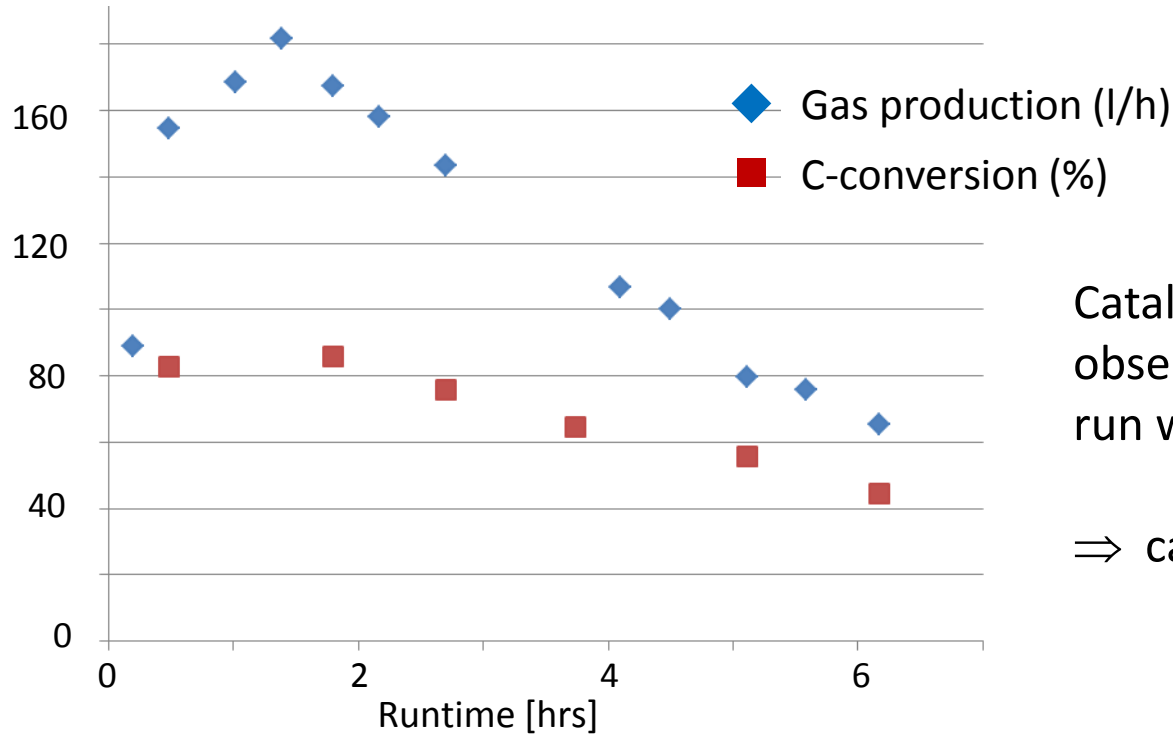
Little improvement with residence time

Sintering and/or deposition of catalyst must be prevented
(research continues in AGATE2)

Continuous SCWG: reactor



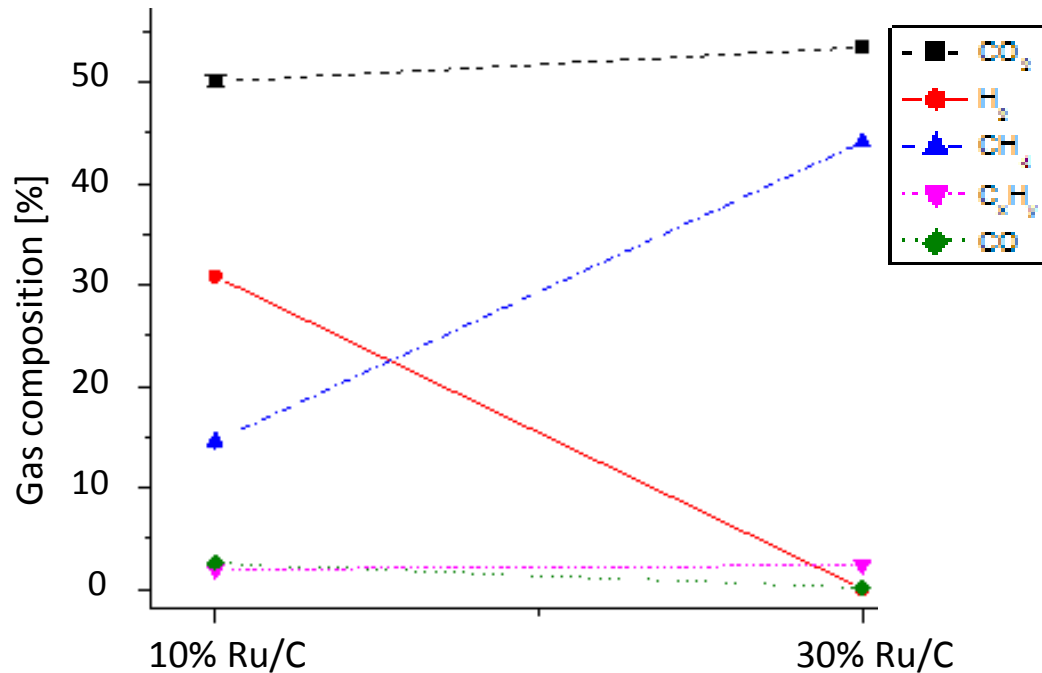
Continuous SCWG: 0.5% Ru/C



Catalytic activity
observed in next test
run without catalyst

⇒ catalyst deposit on
reactor wall

Batch SCWG pyrolysis oil



Carbon to gas conversion lower than for glycerol (10% instead of 80%)

Higher temperature may increase conversion but will reduce CH₄ content

Conclusions of RUG-ST/OC research

80% conversion of glycerol to gas

Gas composition close to equilibrium,

i.e. nearly pure CH_4/CO_2 at 400°C 250 bar

Ru/TiO₂ best performing catalyst

Stability of (nanoparticle) catalysts needs improvement

Conversion of pyrolysis oil more difficult

RUG CIO R&D in AGATE₁

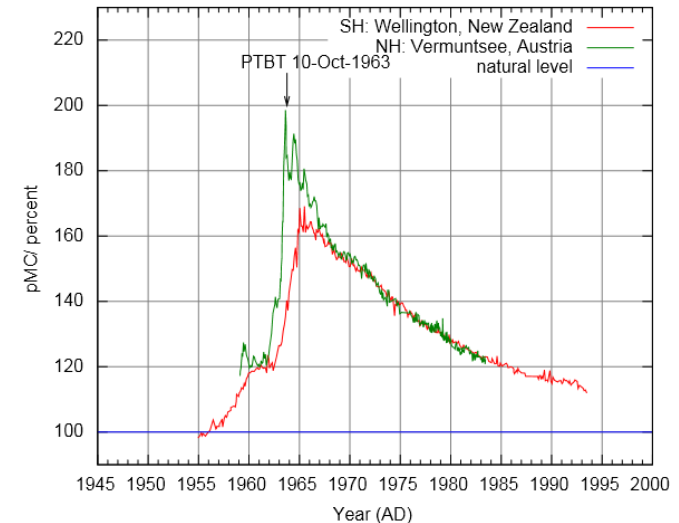
- Develop ¹⁴C analysis method for natural gas, biogas and SNG

WHY and HOW?

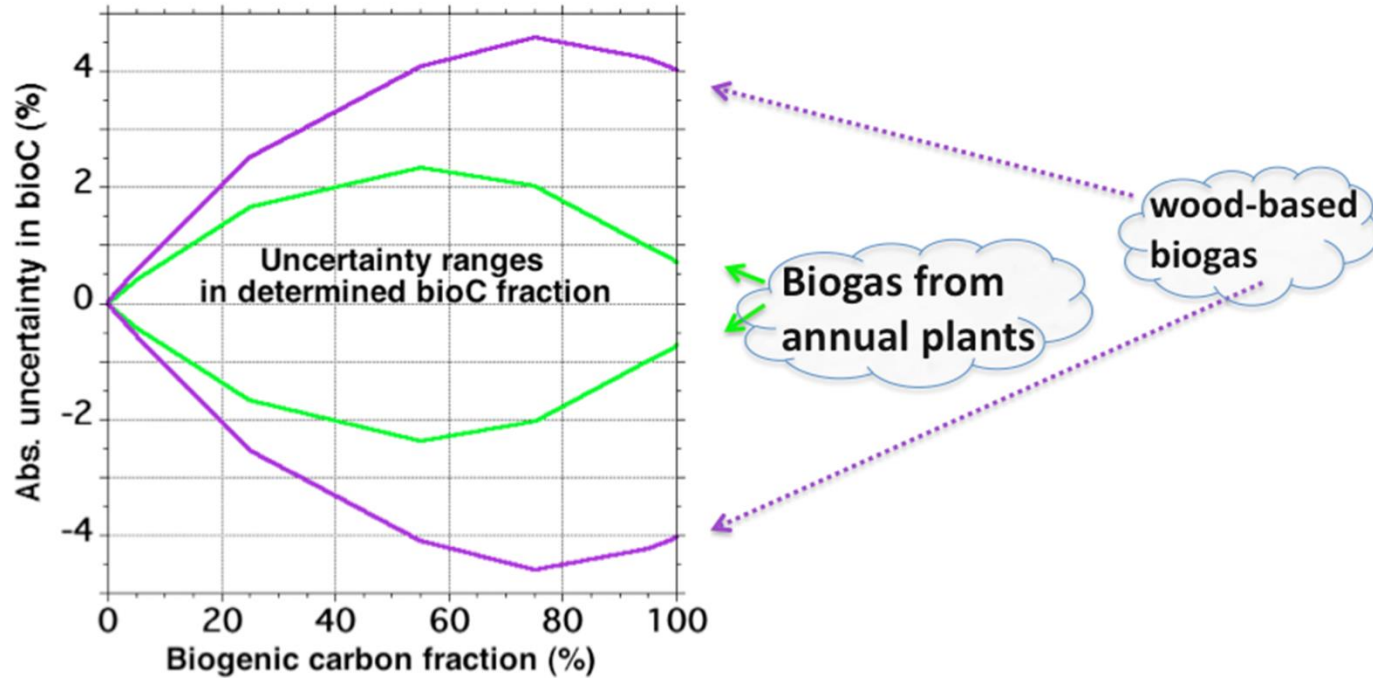
- Allow check of origin (i.e. fossil and/or biomass signature)
- Is already used for waste combustion, based on CO₂ capture
=> combustion of natural gas/biogas/SNG, followed by “standard” analysis

Obstacles in ^{14}C analysis

- ^{14}C content in biomass varies with year of growth
 - Background from nuclear reactions by cosmic rays
 - Peak levels from above-ground nuclear tests
- $^{14}\text{C}/^{12}\text{C}$ disproportionation \Rightarrow compare $^{13}\text{C}/^{12}\text{C}$
- CH_4 and CO_2 in biogas or SNG may have different signatures \Rightarrow separate before combustion
- ECN test rig gases also contain CO , C_2H_4 etc.



Uncertainty in ^{14}C analysis



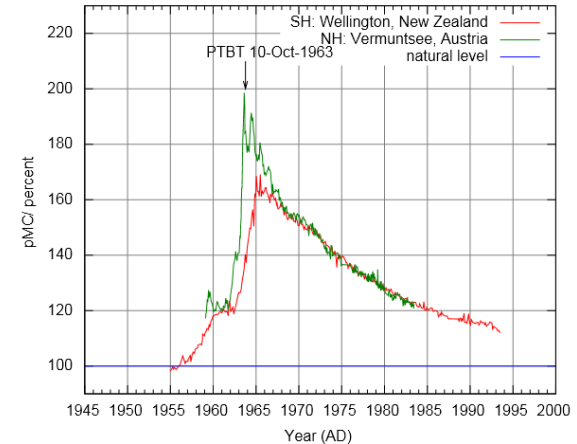
Results (1)

- Quantitative separation, combustion and recovery of C_2H_4 and CO difficult
- Equipment built for separation and combustion of CH_4 (and C_2H_6) from $CH_4/C_2H_6/CO_2$ mixtures
- 3 natural gas and 8 biogas samples analysed

=> good agreement between ^{14}C signals from CO_2 and CH_4/C_2H_6 part:

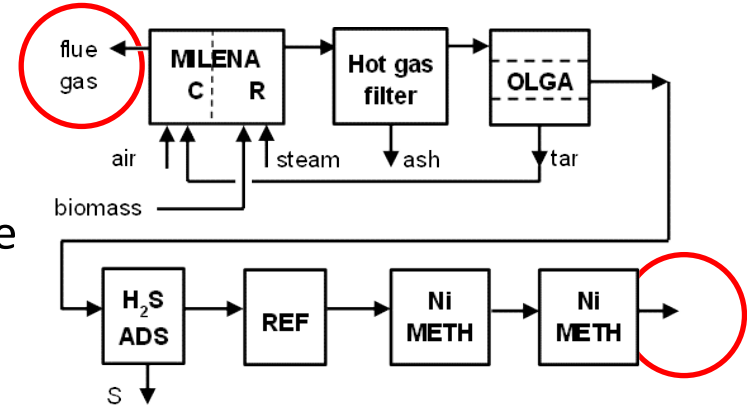
0-1 pMC (% modern carbon) for natural gas, i.e. no ^{14}C

102-105 pMC for biogas, 104 and 116 pMC for two landfill gas samples



Results (2)

- Flue gas & raw SNG from ECN test rig
 - ^{14}C results identical when wood is gasified
 - ^{14}C results different for wood/lignite mixture
- => SNG more biomass signature,
flue gas more fossil signature



Conclusions of RUG-CIO research

^{14}C signature depends on biomass age (also true for waste combustion)

^{14}C signature for biogas from annual crops accurate within a few percent

^{14}C signature for landfill gas requires age correction

^{14}C signature for SNG does reflect fuel signature in case of biomass,
but not if a mixture of fossil fuel and biomass is used

A standard ^{14}C method for biogas and SNG requires further R&D

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Ministerie van Economische Zaken



provincie
groningen

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